

# BBC Biochemical MATERIAL SAFETY DATA SHEET

## Section 1. Chemical Product and Company Information

Common Name:	Hydrochloric Acid 31.5%	Code:	8570, 8573, 8575
Supplier:	BBC Biochemical PO Box 1320 409 Eleanor Lane Mount Vernon, WA 98273 1-800-635-4477	MSDS#:	RW0049
Synonym:	Muratic Acid, Aqueous Hydrogen Chloride, HCl	Validation Date:	3-27-09
Trade Name:	Not available	Print Date:	3-27-09
Material Uses:	Chemical Processing/Metal Cleaning	Responsible Name:	Dr. B
Manufacturer:	Univar USA Inc. 6100 Carillon Point Kirkland, WA 98033 425-889-3400	<b>In Case of Emergency: 1-800-424-9300 Chemtrec USA 1-202-483-7616 Chemtrec Intrl 1-800-635-4477</b>	

## Section 2. Composition and Information on Ingredients

Name	CAS#	% by Weight	OSHA PEL	ACGIH TWA
1) Water	7732-18-5	64-91	Not established	
2) Hydrochloric Acid	7647-01-0	9-36	5ppm	

## Section 3. Hazards Identification

<b>Physical State and Appearance</b>	Clear to light amber liquid with a pungent odor.
<b>Emergency Overview</b>	Harmful by inhalation. Harmful if swallowed. Causes severe burns. Possible sensitizer. Target organs: Liver, Kidneys.
<b>Routes of Entry</b>	Not available.
<b>Potential Acute Health Effects</b>	
<b>Eyes</b>	Corrosive liquid is will cause severe damage to the eyes. High vapor concentrations may also be irritating.
<b>Skin</b>	Corrosive liquid will cause irritation with itching and local redness. Blisters and tissue destruction may occur.
<b>Inhalation</b>	Corrosive liquid will cause irritation and inflammation to mucous membranes. Dizziness may occur.
<b>Ingestion</b>	Corrosive liquid is toxic. May cause headaches, dizziness, nausea and general weakness. Corrosive liquids will cause internal chemical burns in mouth, esophagus and stomach.
<b>Potential Chronic Health Effects</b>	
<b>Medical Conditions Aggravated by Overexposure</b>	Preexisting eye, skin and respiratory disorders may be aggravated by exposure to this product. Impaired function from preexisting disorders may be aggravated by exposure to this product.
<b>Overexposure/Signs/Symptoms</b>	Aspiration pneumonitis may be evidenced by coughing, labored breathing and cyanosis (bluish skin). In severe cases death may result.

## Section 4. First Aid Measures

<b>Eye Contact</b>	Immediately flush eyes with plenty of water for at least 15 minutes while holding eyelids open. Get medical attention.
<b>Skin Contact</b>	Remove contaminated clothing/shoes. Flush skin with water. If irritation occurs, get medical attention. Do not reuse clothing until cleaned.
<b>Inhalation</b>	Remove victim to fresh air. If not breathing, give artificial respiration. If breathing is difficult, oxygen should be administered by qualified personnel. Get medical attention.
<b>Ingestion</b>	Do not induce vomiting. If vomiting occurs spontaneously, keep head below hips to prevent aspiration of liquid into the lungs. Get medical attention.
<b>Notes to Physician</b>	Not available.

## Section 5. Fire Fighting Measures

<b>Flammability of the Product</b>	Not flammable.
<b>Auto-ignition Temperature</b>	Not available
<b>Flash Points</b>	Not available
<b>Flammable Limits</b>	Not available

<b>Products of Combustion</b>	Not available.
<b>Fire Hazards in Presence Of Various Substances</b>	Not available.
<b>Explosion Hazards in Presence of Various Substances</b>	Not available.
<b>Fire Fighting Media and Instructions</b>	Use water fog, "alcohol" foam, dry chemical, or CO <sub>2</sub> . Clear fire area of unprotected personnel. Do not enter confined fire space without full bunker gear, including positive pressure NIOSH approved SCBA. Cool fire exposed containers with water.
<b>Protective Clothing (Fire)</b>	
<b>Special Remarks on Fire Hazards</b>	Fine mist or spray may be flammable at temperatures below the flash point. Carbon monoxide and unidentified organic compounds may be formed during combustion.
<b>Special Remarks on Explosion Hazards</b>	At high temperatures this product can decompose to give off hydrochloric acid and gas.

### Section 6. Accidental Release Measures

<b>Small Spill and Leak</b>	Ventilate area of leak or spill. Remove all sources of ignition. Clean-up personnel require protective clothing and respiratory protection from vapors. Only specially trained or qualified personnel should handle the emergency. Soak up spill and neutralize with soda ash, place in closed containers, label and store in a safe place outdoors to await proper disposal. Spills on areas other than pavement, e.g. dirt or sand, may be handled by removing the affected soils and placing in approved containers.
<b>Large Spill and Leak</b>	Large spills should be removed by a vacuum truck.

### Section 7, Handling and Storage

<b>Handling</b>	Empty containers retain product residue and can be dangerous. Do not pressurize, cut weld, braze, solder, drill, grind, or expose such containers to heat, flame, sparks, static electricity, or other sources of ignition.
<b>Storage</b>	Keep away from heat, sparks, and flame. Surfaces that are hot may ignite liquid product even in the absence of sparks or flame. Store in a cool, ventilated area away from incompatible materials.

### Section 8. Exposure Controls / Personal Protection

<b>Engineering Controls</b>	Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Provide exhaust ventilation sufficient to keep the airborne concentration of this product below its exposure limits. Exhaust air may need to be cleaned by scrubbers or filters to reduce environmental contamination.  Do not add water directly to this product and do not mix with alkalis such as sodium hydroxide (caustic soda) or alkali metals to avoid possible violent reactions. This product may be added to water with mixing and dilution.
<b>Personal Protection</b>	
<b>Eyes</b>	Use chemical safety goggles and/or full face shield where splashing is possible. Contact lenses should not be worn when working with this material. Maintain eye wash fountain and quick-drench facilities in work areas.
<b>Body</b>	Where splashing is possible, full chemically resistant protective clothing (e.g., acid suit) and boots are required.
<b>Respiratory</b>	If exposure may or does exceed occupational exposure limits (Sec. 2) use a NIOSH approved respirator to prevent overexposure. In accord with 29 CFR 1910.134 use either an atmosphere-supplying respirator or an air-purifying respirator for organic vapors.
<b>Hands</b>	Test data indicate the best protection is provided by neoprene, nitrile, and natural rubber gloves.
<b>Feet</b>	Impervious footwear is required in areas where splashing is possible.
<b>Personal Protection in Case of a Large Spill</b>	See above.
<b>Product Name</b>	<b>Exposure Limits</b>
1)Hydrochloric Acid CAS 7647-01-0 2) 3)	OSHA PEL 5ppm TWA
<b>Consult Local authorities before acceptable exposure limits.</b>	

### Section 9. Physical and Chemical Properties

<b>Physical State and Appearance</b>	Clear to light amber liquid with a pungent odor.	<b>Odor:</b>	Pungent
<b>Molecular Weight</b>	36.46	<b>Taste:</b>	Not available
<b>Molecular Formula</b>	HCl	<b>Color:</b>	Clear to light amber.
<b>pH (1%/Water)</b>	Not available		
<b>Boiling/Condensation Point</b>	140 to 221 deg F		
<b>Melting/Freezing Point</b>	-29 to 5 deg F		
<b>Critical Temperature</b>	Not applicable.		

<b>Specific Gravity</b>	1.05 to 1.18
<b>Vapor Pressure</b>	14.6 to 80
<b>Vapor Density</b>	1.3 @ 20 C
<b>Volatility</b>	Not available
<b>Odor Threshold</b>	Not applicable.
<b>Evaporation Rate</b>	Not available
<b>VOC</b>	Not available.
<b>Viscosity</b>	Not available.
<b>Ionicity (in water)</b>	Not available.
<b>Dispersion Properties</b>	Not available.
<b>Solubility</b>	Not available
<b>Physical Chemical Comments</b>	Not available.

### Section 10. Stability and Reactivity

<b>Stability and Reactivity</b>	Stable.
<b>Conditions of Instability</b>	Stable under normal conditions. Avoid heat, flame, and other sources of ignition. Protect from moisture, hygroscopic. Reacts with water, alkalis and metals.
<b>Incompatibility with Various Substances</b>	Avoid strong oxidizing agents, strong bases, metals..
<b>Hazardous Decomposition Products</b>	Carbon monoxide and unidentified organic compounds may be formed during combustion. Generates toxic and irritating gases at high temperatures. Reacts with metals with the evolution of hydrogen which when mixed in air may result in fire or explosion if ignited. Chlorine gas may be released by mixing with strong oxidizers.
<b>Hazardous Polymerization</b>	Will not occur.

### Section 11. Toxicological Information

<b>Toxicity to Animals</b>	CAS 7647-01-0 Hydrochloric Acid LD50: oral, rabbit: 900 mg/kg LC50: inhalation rat: 3124 ppm, 1 hour.
<b>Chronic Effects on Humans</b>	Not available.
<b>Other Toxic Effects on Humans</b>	
<b>Special Remarks on Toxicity to Animals</b>	Not available.
<b>Special Remarks on Chronic Effects on Humans</b>	Not available.
<b>Special Remarks on Other Toxic Effects on Humans</b>	Not available.

### Section 12. Ecological Information

<b>Ecotoxicity</b>	CAS 7647-01-0 Hydrochloric Acid Fish: LC50 (96 Hr) Mosquito Fish: 282 mg/L LC100(24Hr) Trout: 10 mg/L Invertebrates: LC50(48Hr) Starfish: 100-330 mg/L LC50 (48Hr) Shrimp: 100-330 mg/L
<b>BODS and COD</b>	Not available.
<b>Biodegradable/OEDC Mobility</b>	Not available.
<b>Toxicity of the Products of Biodegradation</b>	Not available.
<b>Special Remarks on The Products of Biodegradation</b>	Hydrochloric Acid can be acutely toxic in aquatic life through reduction in aqueous pH to toxic levels. Typically most aquatic species are intolerant of pH levels lower than 5.5 for any extended length of time. Reduction in aqueous pH levels may also cause the liberation of metals such as aluminum which will also contribute to exhibited toxicity. Hydrochloric acid will dissociate in water and undergo neutralization with carbonate and other naturally occurring buffering agents. Terrestrial organisms would be subject to severe burns if exposed to HCl during an accidental release. A large HCl release could lead to a persistent reduction in pH in a poorly buffered system lacking in carbonates or other naturally occurring acid neutralizers. Care should be taken to avoid accidental release to aquatic or terrestrial ecosystems.

**Section 13. Disposal Considerations**

<b>Waste Information</b>	The materials resulting from clean up operations may be hazardous wastes and is therefore subject to specific regulations.
<b>Waste Stream</b>	Not available.
<b>Consult your local or regional authorities.</b>	

**Section 14. Transport Information**

<b>DOT Classification</b>	Hydrchloric acid solution; Class 8; UN1789, Packing Group II
<b>Marine Pollutant</b>	Not available.
<b>Hazardous Substances Reportable Quantity</b>	Not available.
<b>Special Provisions for Transport</b>	Not applicable.
<b>TDG Classification</b>	Hydrchloric acid solution; Class 8; UN1789, Packing Group II
<b>ADR/RID Classification</b>	ADR (Europe) Information not available.
<b>IMO/IMDG Classification</b>	IMDG Information not available.
<b>ICAO/IATA Classification</b>	See IATA Regulations Hydrchloric acid solution; Class 8; UN1789, Packing Group II

**Section 15. Other Information**

<b>Label requirements</b>			
<b>Hazardous Material Information System (U.S.A.)</b>	<b>Health</b>	3	<b>National Fire Protection Association (U.S.A.)</b>
	<b>Fire Hazard</b>	0	
	<b>Reactivity</b>	2	
	<b>Personal Protection</b>		
<b>References</b>			
<b>Other Special Considerations</b>			
<b>Notice to Reader</b>			
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